

## Chapter 7 Review

(Gen)

Name: \_\_\_\_\_

Write the formula of the following compounds:

Ionic(I) Covalent(C)

- I 1) Potassium nitride  $K_3N$
- I 2) Sodium phosphate  $Na_3PO_4$
- C 3) Tetraphosphorus hexoxide  $P_4O_6$
- C 4) Dinitrogen trioxide  $N_2O_3$
- I 5) Ammonium nitrate  $NH_4NO_3$
- I 6) Titanium (II) sulfate  $TiSO_4$
- I 7) Copper (II) hydroxide  $Cu(OH)_2$
- C 8) Carbon tetraiodide  $CI_4$
- C 9) Sulfur dioxide  $SO_2$
- C 10) Dinitrogen pentoxide  $N_2O_5$
- I 11) Aluminum sulfide  $Al_2S_3$
- I 12) Ammonium carbonate  $(NH_4)_2CO_3$

Write the name of the following compounds:

- I 13)  $AlBr_3$  Aluminum bromide
- I 14)  $SrCl_2$  Strontium chloride
- I 15)  $Co_2(SO_4)_3$  Cobalt III sulfate
- C 16)  $XeF_4$  Xenon tetrafluoride
- I 17)  $Zn_2S$  Zinc I sulfide
- C 18)  $P_2S_3$  Diphosphorus trisulfide
- I 19)  $NH_4Br$  Ammonium bromide
- C 20)  $NO_2$  Nitrogen dioxide
- C 21)  $PCl_3$  Phosphorus trichloride
- C 22)  $N_2S$  Dinitrogen monosulfide
- I 23)  $Ba(OH)_2$  Barium hydroxide
- I 24)  $CaCO_3$  Calcium carbonate
- C 25)  $SiO_2$  Silicon dioxide
- C 26)  $NP$  Nitrogen monophosphide

