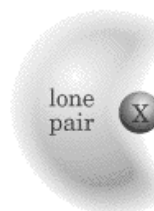


Chapter 8 Notes

Name: _____



Bonds	Lone Pairs	Name	Geometry	Example	Bond Angle
2 atoms		linear		Cl ₂	180°
2	0	linear		CO ₂ or BeCl ₂	180°
3	0	trigonal planar		BF ₃	120°
4	0	tetrahedral		CH ₄	109.5°
3	1	trigonal pyramid		NH ₃	107°
2	2	bent		H ₂ O	104.5°